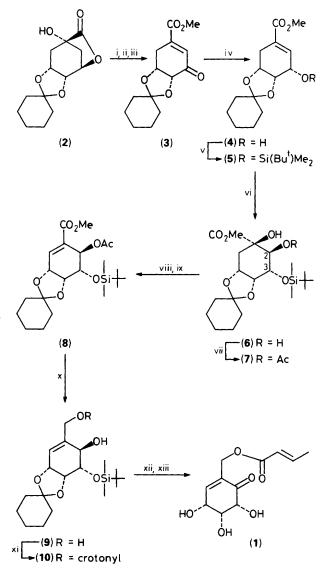
Enantiospecific Synthesis of 2-Crotonyloxy-(4*R*,5*R*,6*R*)-4,5,6-trihydroxycyclohex-2-enone (COTC) from Quinic Acid

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A thirteen-step synthesis of the glyoxalase I inhibitor COTC [2-crotonyloxy-(4*R*,5*R*,6*R*)-4,5,6-trihydroxycyclohex-2-enone] from quinic acid is described.

Recently, the potential of glyoxalase inhibitors as anticancer agents has been indicated.¹ 2-Crotonyloxy-(4R,5R,6R)-4,5,6-trihydroxycyclohex-2-enone (COTC) (1), isolated and charac-



Scheme 1. Reagents and conditions: i, NaOMe/MeOH, 0°C, (96%); ii, dimethyl sulphoxide, oxalyl chloride triethylamine, CH₂Cl₂; iii, POCl₃, pyridine, room temp., (76%); iv, NaBH₄, MeOH, 0°C, (82%); v, Me₂(Bu¹)SiCl, imidazole, *N*,*N*-dimethylaminopyridine (DMAP), CH₂Cl₂, room temp., (96%); vi, OsO₄, trimethylamine-*N*-oxide, Bu¹OH, H₂O, pyridine, reflux, (80%); vii, (MeCO)₂O (Ac₂O), pyridine, DMAP, CH₂Cl₂, (100%); viii, (CF₃SO₂)₂O, pyridine, DMAP, CH₂Cl₂, (86%); ix, triethylamine, 1,8-diazabicylo[54.0]undec-7-ene, CH₂Cl₂, (71%); x, DIBAL-H, tetrahydrofuran, 0°C, (75%); xi, crotonic anhydride, pyridine, DMAP, CH₂Cl₂, (95%); xii, pyridinium chlorochromate, 3 Å molecular sieves, CH₂Cl₂ (80%); xiii, 50% aq. CF₃CO₂H, room temp., (100%).

terised in 1975 as a glyoxalase I inhibitor from cultures of *Streptomyces griseosporeus*,² has been shown to display cytotoxic and cancerostatic activity with low toxicity,³ and to act synergistically with aclarubicin, an anticancer drug.⁴ The absolute configuration of (1) has been confirmed by synthesis.^{5,6} We are interested in its mechanism of tumour inhibition and this communication describes a facile synthesis of COTC (1) *via* a sequence which would afford useful analogues.

The route to COTC (1) is shown in Scheme 1. Adapting the protocol already developed,⁷ the lactone (2),⁸ readily available from quinic acid, was converted into the enone (3), $\dagger m.p.$ 90-91 °C; $[\alpha]_D$ - 44.0° (c 2.1, CH₂Cl₂). Hydride reduction of the keto group in (3) from the less hindered β -face furnished the α -alcohol (4) which was protected as the silvl ether (5), m.p. 54—55 °C; $[\alpha]_D$ + 21.5° (c 2.4, CH₂Cl₂). The double bond in (5) was hydroxylated smoothly to the diol (6), m.p. 97–99 °C; $[\alpha]_D - 18.6^\circ$ (c 1.1, CH₂Cl₂). The stereochemistry of the 2-OH was evident from the ¹H NMR spectrum ($J_{2,3}$ 9.8 Hz). Selective acetylation of (6) gave the monoacetate (7) which was reacted with trifluoromethanesulphonate and underwent base mediated elimination to form the enoate (8), m.p. 82---84 °C; $[\alpha]_D$ - 39.6° (c 0.9, CH₂Cl₂). Di-isobutylaluminium hydride (DIBAL-H) reduction of the diester (8) afforded the diol (9) which was esterified selectively at the primary alcohol to the crotonyl ester (10), $[\alpha]_D = -31.2^\circ$ (c 2.6, CH_2Cl_2). Oxidation of the allylic alcohol (10) followed by hydrolysis furnished COTC (1), m.p. 178–179 °C; $[\alpha]_D$ -106.4° (c 0.6, MeOH) {lit.² m.p. 181 °C; $[\alpha]_{D} - 109^{\circ}$ (c 1.5, MeOH)}.

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[†] All new compounds gave satisfactory analytical and spectral data.